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Correlation of Major Components and Radical Scavenging Activity of Commercial Tea Drinks in Taiwan

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ABSTRACT

Correlation of major components with free radical scavenging activities of 27 commercial tea drinks prepared from green, Oolong, and black teas in Taiwan were investigated. Green tea drinks contained the highest level of total catechins including epigallocatechin gallate (EGCG), epigallocatechin (EGC), epicatechin (EC), epicatechin gallate (ECG), and gallic acid (GA), while these components were low in black tea drinks and moderate in Oolong tea drinks. EGC and EGCG were the major catechins in the three types of tea drinks, and gallic acid showed the highest abundance in black tea drinks. Caffeine concentrations were comparable in the three types of tea drinks. All black tea drinks contained sugars, and sucrose was the main sweetener added. The scavenging abilities against 1,1-diphenyl-2-picrylhydrazyl (DPPH) and 2,2'-azino-bis(3-ethylbenzthiazoline-6-sulphonic acid) (ABTS) radicals were used to determine the antioxidant potential of tea drinks, and the ranked order of potency was green tea > Oolong tea > black tea. Total phenolics, total catechins, and EGC of tea drinks were positively and significantly ($r > 0.8$) correlated to the scavenging abilities against DPPH and ABTS radicals. Our results show that green tea drinks have higher free radical scavenging activity than black and Oolong tea drinks may be related to their high levels of total phenolics and catechins.

Key words: tea drinks, catechins, free radical scavenging, sugar, caffeine

INTRODUCTION

Tea is one of the most widely consumed beverages in Taiwan and many countries throughout the world. Three types of tea are commercially available in Taiwan. They are: green tea (unfermented), Oolong tea (partially fermented) and black tea (fully fermented). They differ in manufacturing conditions, which influence the chemical composition and taste. For the manufacture of green tea, the tea leaves are steamed to inactivate polyphenol oxidase in order to minimize oxidation before further processes. For the manufacture of black tea, the tea leaves are allowed to undergo full oxidation in the fermentation process. Oolong tea is manufactured by allowing partial oxidation of the tea leaves, intermediate between the processes used for the preparation of green and black teas^(1,2). Epidemiological studies suggest that the polyphenolic compounds present in tea leaf have the benefits of reducing the risk of various diseases such as cancer and

cardiovascular diseases^(3,4). The principal hypothesis associated with the putative benefits of tea is linked to the strong free radical-scavenging and antioxidant properties of their polyphenols⁽⁵⁾. Catechins and gallic acid (GA) are considered as the active components of tea polyphenols responsible for the beneficial effects on human health. However, the oral bioavailability of these polyphenols was found to be relatively poor^(6,7). Among various kinds of teas, green tea contains the highest amounts of catechins due to lacking fermentation in the manufacturing process. Four major catechins found in tea are epigallocatechin gallate (EGCG), epigallocatechin (EGC), epicatechin gallate (ECG), and epicatechin (EC). EGCG is the most abundant catechin in tea and is believed to be the most bioactive component⁽⁸⁾. In addition to catechins, tea also contains moderate amount of caffeine (CA), a central nervous system (CNS) stimulant⁽⁹⁾.

Because catechins, GA, and CA contents are related to the quality of tea, the constituents of commercial teas have been the subject of intensive investigation for a long time. However, relatively less attention has been directed to the

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constituents of commercial tea drinks. Since the manufacture processes of the tea drinks, such as amounts of tea used, brewing methods and sterilization, etc., vary, resulting profile of polyphenols or catechins in tea drinks may not be the same as that reported in tea leaves. In addition, supplementation with or without other additives (i.e. flavor, ascorbic acid, sucrose, etc.) in tea drinks would make them more complicated. Due to their convenience, canned and bottled tea drinks are becoming more popular worldwide. The catechins in tea drinks, however, may be reduced during manufacture process and storage. For instance, degradation and/or epimerization of epicatechins by heat treatment may occur in tea drinks^(10,11), and the concentration of GCG present in green tea drinks is likely to increase due to the epimerization of EGCG during autoclaving⁽¹¹⁾. The addition of ascorbic acid to green tea drinks was shown to stabilize catechins⁽¹²⁾. In addition, catechins are labile in alkaline solution. Catechins in green tea at pH values of > 6 are degraded with storage time^(11,13). These results indicate catechins are not always stable during the manufacturing process and storage. Some tea drinks are supplemented with sugars to reduce bitterness of tea and to provide energy source. Recent studies demonstrated that sugar-sweetened beverages are a significant contributor to weight gain and can lead to increased risks of type 2 diabetes mellitus and cardiovascular diseases^(14,15). Therefore, to support the beneficial health effects of tea drinks, more data of tea components should be provided on the labels of commercial tea drinks.

In this study, we evaluated the major components of 27 tea drinks, including green, Oolong, and black tea, commercially available in Taiwan and their correlations to free radical scavenging activities. Free radical scavenging activities of the tea drinks were evaluated by examining the scavenging abilities against the 1,1-diphenyl-2-picrylhydrazyl (DPPH) and 2,2'-azino-bis(3-ethylbenzthiazoline-6-sulphonic acid) (ABTS) radicals. Correlations between the major components including catechins, GA, CA, ascorbic acid, and sugars (sucrose, fructose, and glucose) and free radical scavenging activities of the tea drinks were also determined to verify the contribution of each component to the antioxidant activity.

MATERIALS AND METHODS

I. Chemicals and Reagents

EC, EGC, EGCG, ECG, GA, gallic acid, gallic acid gallate (GCG), and CA, Folin-Ciocalteu reagent, DPPH, ABTS, aluminum chloride hexahydrate, sodium ascorbate, formic acid, and ammonium acetate were all obtained from Sigma-Aldrich (St. Louis, MO, USA). All other chemicals and reagents were of analytical grade and were obtained commercially.

II. Tea Drinks

Twenty-seven commercial bottled tea drinks, including

green, Oolong, and black tea, were purchased from local markets in Taichung. The ingredient labels of the tea drinks are described in Table 1. The tea drinks were filtered with a 0.45- μ m filter, and the filtrate was used to determine the components and their free radical scavenging activities.

III. Determination of the Concentration of Total Phenolics in Tea Drinks

The concentrations of total phenolics in the tea drinks were determined using the Folin-Ciocalteu reagent⁽¹⁶⁾. A tea drink sample (20 μ L) was mixed with 90 μ L of Folin-Ciocalteu reagent. The mixture was shaken vigorously for 3 min, and then 175 μ L of 7.5% Na_2CO_3 (w/v) was added. After incubation at room temperature for 30 min, the absorbance of the reaction mixture was measured at 750 nm. Gallic acid (GA) was chosen as a standard to construct a standard curve. The total phenolics in tea drinks are expressed in μ g of gallic acid equivalents/mL.

IV. Determination of the pH Value

The pH of the tea drinks was measured using a SUNTEX microprocessor pH meter sp-2200 (Taipei, Taiwan).

V. Determination of EGC, EGCG, GCG, EC, ECG, GA, and CA in Tea Drinks

EGC, EGCG, GCG, EC, ECG, GA, and CA concentrations in the tea drinks were determined according to the method of Cabrera *et al.*⁽⁶⁾ with modifications. The filtrate of a tea drink was injected into a Hitachi-L7400 Series LC System (Tokyo, Japan) equipped with an ultraviolet (UV) detector set at 280 nm. An Agilent Extend-C18 reversed-phase column (5 μ m, 250 \times 3.0 mm) was used. The mobile phase consisted of 10 mM ammonium acetate containing 0.5% formic acid (Solvent A) and methanol containing 0.5% formic acid (Solvent B). The flow rate was 0.8 mL/min, and the total running time was 50 min. The gradient system was 2% B (0 - 1 min), 2% B to 43% B (1 - 27 min), 43% B to 90% B (27 - 33 min), 90% B to 2% B (33 - 40 min), and 2% B (40 - 50 min). The column temperature was at ambient temperature (25°C). Target peaks in the chromatograms were identified by comparing the retention times of reference standards. The retention times of GA, EGC, CA, EGCG, EC, GCG, and ECG were 5.0, 15.8, 17.9, 19.4, 20.6, 21.6, and 24.3 min, respectively. The concentrations of these compounds in the tea drinks were determined with the calibration curves of authentic standards. The calibration curves were linear over a concentration range of 0.2 - 300 μ g/mL with correlation coefficients ≥ 0.999 . The accuracy of the method was evaluated from the recovery assays which were calculated from the measured concentration vs. nominal concentration by the following equation:

$$\text{Accuracy (\%)} = (\text{Measured concentration} \div \text{Nominal concentration}) \times 100.$$

Table 1. Product name and ingredients on the labels of the commercial tea drinks examined

Sample number	Product name	Ingredient label
Green tea drink		
1	Kutao Super Green Tea	Fiber
2	Cha Li Won Japanese Style Green Tea Sugar Free	Flavor, vitamin C
3	AGV Health and Oil Trim	Fiber
4	Royal Tea Garden Tea Japanese Sugar-free Green Tea	-
5	Every Morning Green Tea	Fiber, flavor, vitamin C
6	Uni-President Green Tea	Flavor, vitamin C
7	Life Foam Green tea	Sugar, flavor
8	Kuang Chuan Cold Brew Green Tea	Sugar, flavor, vitamin C
9	Y.E.S Health and Oil Trim	Vitamin C
10	Pure Tea Sugar-free Green Tea	Flavor
Black tea drink		
11	Kuang Chuan Cold Brew Black Tea	Sugar, flavor, vitamin C
12	Taisun Iced Black Tea	Sugar, flavor, malic acid
13	Cha Li Won English Tea	Sugar, flavor, vitamin C
14	Life Foam Black Tea	Sugar, flavor
15	Mine Chine Black Tea	Sugar, flavor, vitamin C
16	Uni-President Black Tea	Sugar, flavor, vitamin C
17	Pure Tea Black Tea	Sugar, flavor
Oolong tea drink		
18	Fantastic Tea	Vitamin C
19	Every Morning Black Oolong Tea	Fiber, vitamin C
20	Kuang Chuan Cold Brew Oolong Tea	Sugar, flavor
21	Dong Ding Oolong Tea	Sugar, vitamin C
22	Uni-President Oolong Tea	Flavor, vitamin C
23	Cha Li Won Alishan Oolong Tea	Flavor, vitamin C
24	Royal Tea Garden Tea Cold Mountain Oolong Tea	-
25	Cha Li Won Ching Sing Oolong Tea	Flavor, vitamin C
26	Cha Li Won Pekoe Oolong Tea	Sugar, vitamin C
27	Pure Tea Oolong Tea	Sugar, flavor, vitamin C

The acceptance criterion of accuracy was 95 - 105%. The precision of the method was expressed as the percent (%) relative standard deviation (RSD). The RSD (%) = Standard deviation of the QC group (accuracy) ÷ Average accuracy of the QC group × 100. The acceptance criterion was ≤ 10%. The limit of detection (LOD) and limit of quantification (LOQ) were calculated as the concentration giving a signal equal to 3× and 10× the signal/noise (S/N) ratio.

VI. Determination of Ascorbic Acid

Ascorbic acid concentration in the tea drinks was determined following the method published by de Quirós *et al.*⁽¹⁷⁾. Sample preparations and the mobile phase solvents used were the same as those used for the determination of catechins in the tea drinks. A Supelco C18 column (5 µm, 250 × 4.6 mm)

was used. The gradient system was: 10% B (0 - 3 min), 10% B to 90% B (3 - 4 min), 90% B (4 - 6 min), 90% B to 10% B (6 - 7 min), and 10% B (7 - 16 min). The effluent was monitored using a UV detector set at 245 nm. The flow rate was 0.8 mL/min, and the total running time was 16 min. The retention time of ascorbic acid was 3.1 min. The concentration of the calibration curve ranged 12.5 - 500 µg/mL with correlation coefficients ≥ 0.999. The concentration of ascorbic acid in the tea drinks was calculated by comparing the peak area of ascorbic acid in the drinks with that of the standard curve.

VII. Determination of Sucrose, Fructose, and Glucose Contents

Sucrose, fructose, and glucose concentrations in the tea

drinks were determined by an enzymatic method according to the instruction of the assay kit from the manufacturer (Megazyme, Bray, Ireland).

VIII. Determination of the Scavenging Abilities of the Tea Drinks Against the DPPH and ABTS Radicals

Tea drink samples were diluted 200 and 400 folds, respectively, with deionized water prior to being analyzed for the scavenging abilities against DPPH and ABTS radicals.

The scavenging activity of tea drinks against the DPPH free radical was determined by the method of Zhao *et al.*⁽¹⁸⁾ with some modifications. An aliquot (224 μ L) of diluted tea drink sample was added to 56 μ L of an 80 μ M DPPH solution in methanol. The mixture was then shaken vigorously and allowed to stand at room temperature for 30 min. Deionized water (224 μ L) only was used as the experimental control. The absorbance of the reaction mixture was measured at 517 nm.

The scavenging activity of tea drinks against the ABTS free radical was determined by the method of Ozgen *et al.*⁽¹⁹⁾ with some modifications. The ABTS free radical solution was prepared by dissolving 7 mM of ABTS and 2.45 mM of potassium persulfate in a 20 mM acetic acid solution. This mixture was allowed to stand for 16 h at room temperature in the dark until reaching a stable oxidative state. The ABTS solution was then diluted with 20 mM acetic acid solution (pH 4.5) to obtain a solution with an absorbance of 0.700 ± 0.01 at 734 nm. The ABTS solution (20 μ L) was added to 230 μ L of the tea drink sample solution. Deionized water (230 μ L) only was used as the experimental control. The mixture was then shaken vigorously and allowed to stand at room temperature for 30 min. The absorbance of the reaction mixture was measured at 734 nm.

The percent inhibition of the DPPH and ABTS radicals was calculated according to the following formula:

$$\text{Inhibition (\%)} = \frac{(\text{Absorbance of control} - \text{Absorbance of test sample})}{\text{Absorbance of control}} \times 100.$$

Effects of the components of tea drinks including ascorbic acid, GA, EGCG, and GCG on the inhibition of DPPH and ABTS radicals were evaluated, and the concentration (μ g/mL) of these components required to deplete the amount of both radicals by 50% (IC_{50} value) were determined graphically. The maximum inhibitory actions of the scavenging abilities against the DPPH and ABTS radicals by tea drinks were 82% and 95%, respectively.

IX. Data Analysis

All experiments were performed, at least, in triplicate. Data are presented as the mean \pm SD. A linear regression analysis was performed, and the correlation coefficient, *r*, between the scavenging abilities of the free radicals (DPPH and ABTS) and tea components was determined.

RESULTS AND DISCUSSION

I. The Contents of Total Phenolics and the Scavenging Abilities Against the DPPH and ABTS Radicals in Tea Drinks

The contents of total phenolics in the 27 commercial tea drinks are shown in Table 2. The contents of total phenolics in tea drinks ranged 130 - 639 μ g/mL. The highest level was found in a green tea drink (639 μ g/mL; Sample 3). The ranked order of mean values of total phenolics in the three types of tea drinks were green tea (485 ± 94 μ g/mL) > Oolong tea (348 ± 14 μ g/mL) > black tea (236 ± 70 μ g/mL). The scavenging abilities of the tea drinks on DPPH and ABTS radicals were used to determine their antioxidant activities^(18,19). The percent inhibition against DPPH and ABTS free radicals generation by each tea drink is shown in Table 2. Due to the high scavenging activity of the tea drinks against the DPPH and ABTS free radicals, all tea drinks were diluted 200 and 400 folds for the DPPH and ABTS assays, respectively. The scavenging activity of the 27 tea drinks against DPPH and ABTS free radicals ranged 24.2 - 70.6% and 23.0 - 91.7%, respectively. As expected, the green tea drinks showed the highest scavenging activity against DPPH ($57.4 \pm 7.9\%$) and ABTS ($75.3 \pm 10.0\%$) free radicals followed by Oolong tea drinks (DPPH: $45.6 \pm 9.7\%$; ABTS: $62.9 \pm 13.5\%$). Black tea drinks were found to have the lowest scavenging activity (DPPH: $28.9 \pm 4.6\%$; ABTS: $38.7 \pm 11.2\%$). The higher antioxidant activity observed in green tea drinks is probably due to the higher content of total phenolics in green tea. Catechins, the major components of polyphenols, are probably the major contributor to the antioxidant activity of green tea drinks^(1,2).

II. Concentrations of Catechins (EGC, EGCG, GCG, EC, ECG), GA, and CA in Tea Drinks

Figure 1 shows the HPLC-UV chromatograms of reference standards and representative samples of green (Sample No. 10), Oolong (Sample No. 17) and black (Sample No. 27) tea drinks. Table 3 summarizes the analytical data of the standard curves. Table 4 shows the pH and the mean \pm SD concentrations of EGC, EGCG, GCG, EC, ECG, GA, CA and ascorbic acid in each commercial tea drink. Green tea drinks were found to have the highest concentrations of catechins, which was followed by Oolong tea drinks. Black tea drinks had the lowest concentrations of catechins. This observation is in agreement with the degree of fermentation of tea during the manufacturing process^(1,2). EGC was identified as the most abundant catechin in all tea drinks except for Samples 10, 17, and 27. The ranked order of the mean values of each catechin in the three types of tea drinks was EGC > EGCG > GCG > ECG = EC. This result is inconsistent with a previous study that EGCG was found to be the most abundant catechin in green tea⁽²⁾ as well as in green tea drinks⁽²⁰⁾. Tsai *et al.*⁽²¹⁾ reported relatively close values on a dry-weight basis (% w/w) for EGCG ($5.1 \pm 2.3\%$) and EGC ($4.5 \pm 1.5\%$)

Table 2. Total phenolics and the antioxidant activities of 27 commercial tea drinks examined in this study ^a

	Total phenolics (µg/mL)	Inhibition of free radicals	
		DPPH (%)	ABTS (%)
Green tea			
1	383 ± 35	58.6 ± 0.9	68.9 ± 0.5
2	491 ± 57	53.3 ± 2.4	71.9 ± 3.2
3	639 ± 69	67.2 ± 4.8	91.7 ± 0.8
4	345 ± 60	45.8 ± 3.1	64.9 ± 0.1
5	596 ± 85	70.6 ± 1.9	90.2 ± 1.6
6	452 ± 79	53.0 ± 1.8	70.8 ± 1.7
7	437 ± 40	51.8 ± 2.6	72.0 ± 1.0
8	467 ± 10	50.9 ± 1.3	62.0 ± 4.7
9	466 ± 8	60.4 ± 1.7	79.0 ± 0.3
10	577 ± 68	62.9 ± 1.2	81.3 ± 0.3
Mean ± SD	485 ± 94	57.4 ± 7.9	75.3 ± 10.0
Black tea			
11	180 ± 13	28.1 ± 4.4	28.6 ± 0.4
12	130 ± 0.2	24.7 ± 1.3	23.0 ± 0.1
13	315 ± 13	57.2 ± 5.7	56.3 ± 0.5
14	234 ± 9	26.8 ± 3.8	40.6 ± 0.5
15	262 ± 16	34.0 ± 1.3	42.2 ± 0.9
16	211 ± 13	24.2 ± 1.6	34.5 ± 1.1
17	323 ± 13	33.2 ± 2.9	45.9 ± 0.5
Mean ± SD	236 ± 70	28.9 ± 4.6	38.7 ± 11.2
Oolong tea			
18	373 ± 14	50.9 ± 3.4	67.7 ± 0.5
19	467 ± 16	52.8 ± 2.7	74.3 ± 1.2
20	324 ± 29	42.6 ± 1.5	59.0 ± 1.1
21	242 ± 16	39.4 ± 1.8	49.1 ± 0.2
22	233 ± 6	30.9 ± 2.5	42.6 ± 0.3
23	480 ± 17	59.1 ± 2.4	80.9 ± 0.1
24	342 ± 26	51.6 ± 1.4	62.7 ± 0.5
25	372 ± 7	50.4 ± 0.5	69.6 ± 0.2
26	273 ± 3	32.5 ± 2.9	45.9 ± 0.5
27	369 ± 53	58.1 ± 3.5	77.2 ± 0.4
Mean ± SD	348 ± 14	45.6 ± 9.7	62.9 ± 13.5
Reference standards ^b			
Ascorbic acid (5 µg/mL)		37.4 ± 1.6	42.2 ± 0.7
Gallic acid (5 µg/mL)		70.7 ± 1.8	80.1 ± 0.7
EGCG (5 µg/mL)		73.7 ± 1.2	79.7 ± 0.3
GCG (5 µg/mL)		74.3 ± 2.4	87.7 ± 0.2

^a Tea drinks were respectively diluted 200 and 400-folds with deionized water for the DPPH and ABTS assays, respectively. Data are expressed as the mean ± standard derivation (n = 3).

^b IC₅₀ values of ascorbic acid, gallic acid (GA), epigallocatechin gallate (EGCG), and gallic acid gallate (GCG) for the DPPH assay were 4.9, 2.5, 2.3, and 2.2 µg/mL, respectively. The corresponding values of ascorbic acid, GA, EGCG, and GCG for the ABTS assay were 5.9, 2.1, 2.4, and 2.2 µg/mL, respectively.

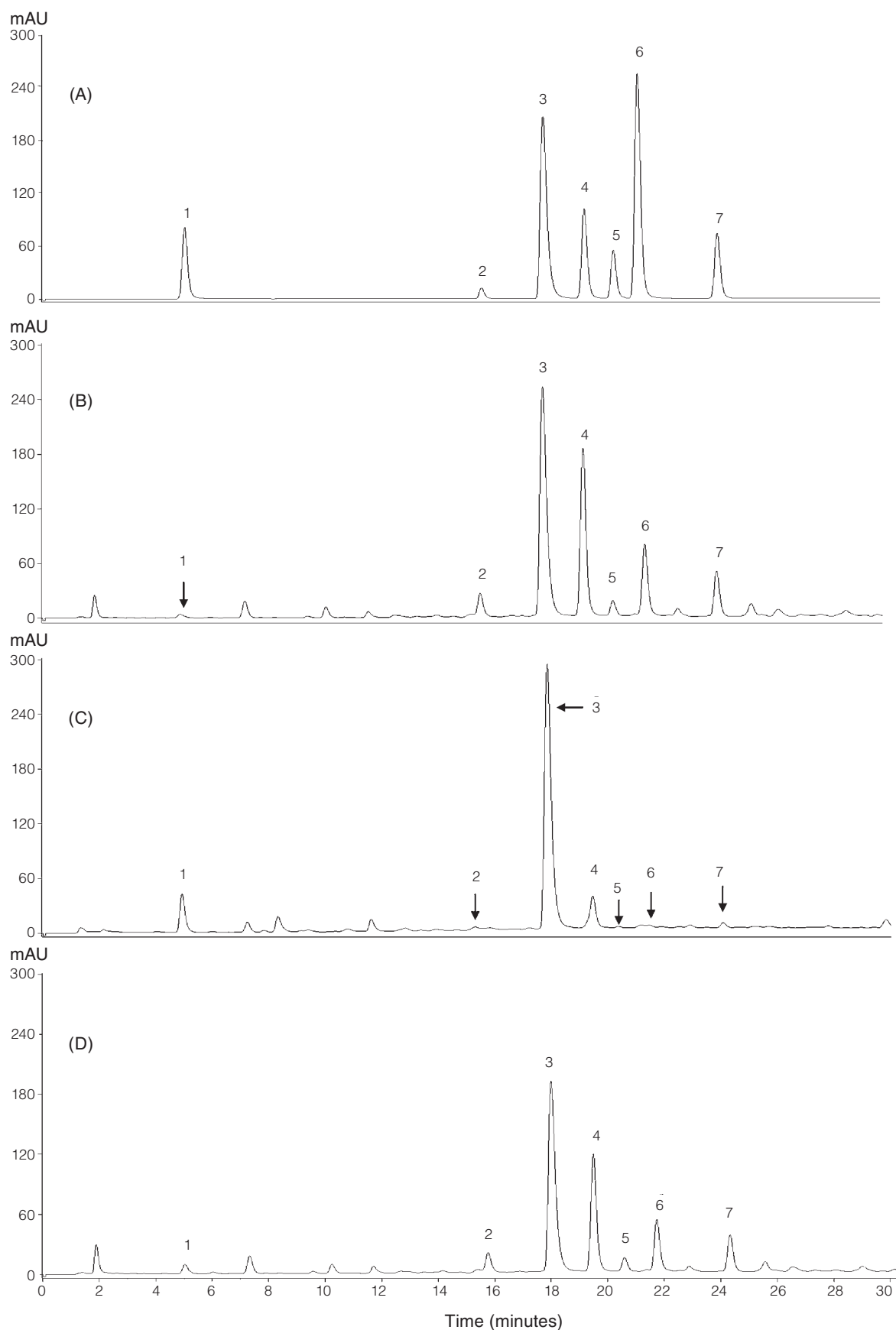


Figure 1. HPLC-UV chromatograms of (A) standards, (B) green tea drink, (C) black tea drink and (D) Oolong tea drink at 280 nm. Peaks: 1, gallic acid (GA); 2, epigallocatechin (EGC); 3, caffeine (CA); 4, epigallocatechin gallate (EGCG); 5, epicatechin (EC); 6, galocatechin gallate (GCG); 7, epicatechin gallate (ECG).

Table 3. Calibration curve equations, correlation coefficients, limit of detection (LOD), limit of quantification (LOQ), reproducibility (RSD in %), and recovery of seven tea drink components examined in this study

Component	Calibration curve equations (n = 4)	Correlation Coefficient (n = 4)	Linear range ($\mu\text{g/mL}$) (n = 4)	LOD ($\mu\text{g/mL}$) (n = 4)	LOQ ($\mu\text{g/mL}$) (n = 4)	RSD (%) (n = 10)	Recovery (%) (n = 4)
GA	$y = (0.0223 \pm 0.0002)x - (4.9503 \pm 0.4678)$	0.9998	0.2 - 100	0.0125	0.05	0.6	99
EGC	$y = (0.0015 \pm 0.0000)x - (0.3932 \pm 0.3162)$	0.9999	0.4 - 250	0.2	0.4	1.0	108
CA	$y = (0.0234 \pm 0.0004)x - (10.2826 \pm 7.5065)$	0.9999	0.6 - 200	0.0375	0.15	0.4	101
EGCG	$y = (0.0092 \pm 0.0002)x - (12.3443 \pm 1.1374)$	0.9992	0.6 - 300	0.075	0.15	0.5	105
EC	$y = (0.007 \pm 0.0001)x - (1.4979 \pm 0.9939)$	0.9999	0.4 - 100	0.1	0.4	0.6	99
ECG	$y = (0.0209 \pm 0.0003)x - (9.6684 \pm 1.2561)$	0.9990	0.2 - 50	0.025	0.05	0.5	101
GCG	$y = (0.0209 \pm 0.002)x - (13.5156 \pm 7.1249)$	0.9999	0.2 - 200	0.1	0.2	0.7	110

GA, gallic acid; EGC, epigallocatechin; CA, caffeine; EGCG, epigallocatechin gallate; EC, epicatechin; ECG, epicatechin gallate; GCG, gallocatechin gallate.

in different varieties of commercial green teas in Taiwan. Therefore, the cultivated variety of teas rich in EGC may be a contributing factor for the high concentration of EGC observed in the commercial tea drinks in Taiwan. It has been reported that the GCG concentration of total catechins could increase from approximately < 1.5% of normal level to as much as 50% in some tea drinks⁽¹¹⁾. However, in this study, a moderate amount of GCG ($53.6 \pm 21.2 \mu\text{g/mL}$; accounting approximately 14.5% of total catechins) was observed in the green tea drinks. The high GCG concentration found in the green tea drinks is probably derived from the thermal conversion of EGCG during autoclaving^(11,22).

The stability of catechins in tea drinks varies with the ingredients and pH of the tea drinks^(11,12). Sucrose appears to have no effect on catechins but ascorbic acid exerts a protective effect from degradation of catechins⁽¹¹⁾. The stability of catechins in tea drinks is known to be sensitive to pH⁽¹²⁾. At pH 4, catechins were found stable for 28 days at 25°C ⁽¹³⁾ while being stable only for 20 min at 120°C ⁽¹¹⁾. Catechins were also found to degrade gradually or underwent epimerization at pH 6. In this study, green tea drinks except Sample 1 had pH > 6 (mean value: 6.3 ± 0.3). The concentrations of ascorbic acid in green tea drinks vary widely, and it was higher than those in the Oolong and black tea drinks (Table 4). It was shown that supplementation of ascorbic acid to a concentration of 0.2 mg/mL only cause little or no change in the pH of the tea drink, but the stability of catechins can significantly be improved⁽¹²⁾. Thus, the higher ascorbic acid supplemented in green tea drinks than that in Oolong or black tea drinks may reduce the degradation of catechins during storage.

In this study, GA was detected in all tested tea drinks ($2.4 - 64.4 \mu\text{g/mL}$). Black tea drinks were found to have the highest concentration of GA ($24.9 \pm 18.4 \mu\text{g/mL}$). This result was accompanied by notably lower pH values of black tea drinks (5.5 ± 0.9) than those of green (6.3 ± 0.3) and Oolong (6.3 ± 0.2) tea drinks. Fernández *et al.*⁽²³⁾ analyzed the contents of GA in 45 tea samples and found higher GA and lower catechin levels in fermented teas. Zuo *et al.*⁽²⁾ reported that the fermentation process enhanced the liberation of GA from galloylcatechins in green tea and that both Pu-erh and black teas contain remarkably high level of GA. Since the pKa of GA is at 4.4, which may result in proton dissociation of the carboxyl group of GA from the galloyl moiety of some catechins in tea drinks, it is thus suggested that the lower pH observed in black tea drinks might possibly be related to the higher GA level.

Concentrations of CA in all tea drinks varied ($49.8 - 205 \mu\text{g/mL}$) but the mean concentrations among the three types of tea drinks are comparable (green tea: $137 \pm 29 \mu\text{g/mL}$; black tea: $136 \pm 50 \mu\text{g/mL}$; and Oolong tea: $144 \pm 41 \mu\text{g/mL}$). Fernández *et al.*⁽²³⁾ and Cabrera *et al.*⁽⁶⁾ reported that CA was higher in black tea than green and Oolong teas. The reasons why CA was higher in black teas and comparable among the three types of tea drinks were unknown. Since the manufacture processes of the tea drinks, such as amounts of tea used, brewing methods and sterilization, etc., may vary, resulting profile of CA and catechins in tea drinks may not

Table 4. Concentrations ($\mu\text{g/mL}$) of catechins, gallic acid (GA), caffeine (CA), ascorbic acid and pH of tea drinks examined in this study

	pH	GA	EGC	CA	EGCG	GCG	EC	ECG	Total catechins	Ascorbic acid
Green tea drink										
1	5.8	5.6 \pm 0.6	122 \pm 9	108 \pm 8	86.6 \pm 6.2	43.6 \pm 2.9	10.9 \pm 1.0	9.1 \pm 0.5	273 \pm 20	204 \pm 10
2	6.7	8.8 \pm 0.3	177 \pm 3	118 \pm 4	94.6 \pm 3.0	47.5 \pm 1.0	15.1 \pm 0.3	12.6 \pm 0.3	347 \pm 7	46.8 \pm 40.2
3	6.1	7.9 \pm 0.1	186 \pm 3	181 \pm 2	142 \pm 1	103 \pm 1	16.7 \pm 0.1	23.2 \pm 1.1	471 \pm 4	484 \pm 33.9
4	6.3	4.7 \pm 0.1	113 \pm 1	95.8 \pm 1.2	90.0 \pm 0.7	42.9 \pm 0.3	11.9 \pm 0.2	9.9 \pm 0.0	267 \pm 1	223 \pm 9
5	6.6	6.5 \pm 0.1	214 \pm 8	173 \pm 6	165 \pm 6	78.3 \pm 2.6	22.0 \pm 0.6	17.5 \pm 0.5	497 \pm 18	251 \pm 24
6	6.3	9.8 \pm 0.2	165 \pm 2	125 \pm 3	93.3 \pm 1.2	45.4 \pm 0.8	13.8 \pm 0.4	13.1 \pm 0.2	331 \pm 4	98.1 \pm 27.2
7	6.1	8.0 \pm 0.7	167 \pm 1	128 \pm 1	97.7 \pm 0.4	53.4 \pm 0.8	15.3 \pm 0.1	18.2 \pm 0.1	352 \pm 1	156 \pm 9
8	6.3	10.6 \pm 0.7	152 \pm 1	131 \pm 2	40.7 \pm 5.6	29.7 \pm 1.5	19.5 \pm 0.2	21.6 \pm 0.3	263 \pm 7	4.5 \pm 3.7
9	6.0	21.1 \pm 0.4	162 \pm 0	140 \pm 4	131 \pm 1	43.1 \pm 1.1	22.8 \pm 0.4	14.1 \pm 0.3	372 \pm 3	99.7 \pm 6.4
10	6.6	2.4 \pm 0.1	173 \pm 17	173 \pm 9	278 \pm 38	49.0 \pm 0.7	26.1 \pm 1.0	35.5 \pm 1.2	512 \pm 56	21.1 \pm 17.1
Mean \pm SD	6.3 \pm 0.3	8.6 \pm 5.0	163 \pm 29	137 \pm 29	122 \pm 65	53.6 \pm 21.2	17.4 \pm 5.0	17.5 \pm 7.9	369 \pm 95	159 \pm 143
Black tea drink										
11	5.9	16.5 \pm 0.5	23.7 \pm 0.3	152 \pm 7	2.1 \pm 0.0	1.5 \pm 0.2	4.0 \pm 0.1	1.2 \pm 0.1	32.0 \pm 0.5	6.2 \pm 2.0
12	3.4	7.2 \pm 0.3	6.7 \pm 0.8	49.8 \pm 2.0	3.0 \pm 0.1	0.1 \pm 0.0	1.4 \pm 0.0	1.8 \pm 0.1	13.0 \pm 0.9	58.2 \pm 11.8
13	5.9	64.4 \pm 1.7	55.0 \pm 6.6	131 \pm 3	BLQ	2.6 \pm 0.3	12.7 \pm 0.3	0.7 \pm 0.2	70.2 \pm 7.9	6.8 \pm 2.7
14	5.8	22.1 \pm 1.2	21.1 \pm 2.4	116 \pm 7	3.5 \pm 0.1	0.9 \pm 0.0	2.6 \pm 0.3	1.7 \pm 0.1	29.5 \pm 2.5	4.7 \pm 3.2
15	5.8	24.5 \pm 1.3	33.7 \pm 3.7	181 \pm 9	2.0 \pm 0.1	1.8 \pm 0.0	5.1 \pm 0.6	2.2 \pm 0.0	44.2 \pm 3.2	4.4 \pm 3.2
16	6.1	16.4 \pm 0.5	16.3 \pm 1.6	120 \pm 3	1.7 \pm 0.1	0.9 \pm 0.1	2.5 \pm 0.2	1.5 \pm 0.1	22.7 \pm 1.4	4.3 \pm 1.6
17	5.7	23.5 \pm 0.9	26.8 \pm 1.5	205 \pm 14	47.0 \pm 2.0	1.8 \pm 0.1	3.0 \pm 0.2	3.1 \pm 0.2	81.1 \pm 2.3	6.0 \pm 0.9
Mean \pm SD	5.5 \pm 0.9	24.9 \pm 18.4	26.2 \pm 15.3	136 \pm 50.2	9.9 \pm 18.2	1.4 \pm 0.8	4.5 \pm 3.8	1.8 \pm 0.8	41.8 \pm 25.2	12.9 \pm 20.0
Oolong tea drink										
18	6.1	4.3 \pm 0.2	122 \pm 6	116 \pm 5	112 \pm 7	48.4 \pm 2.8	13.4 \pm 0.9	13.7 \pm 0.7	293 \pm 32	59.4 \pm 6.8
19	6.5	16.3 \pm 1.6	113 \pm 16	205 \pm 2	105 \pm 2	49.1 \pm 0.4	13.2 \pm 1.4	12.7 \pm 0.8	242 \pm 107	5.0 \pm 2.4
20	6.6	5.5 \pm 1.1	71.8 \pm 15.2	78.0 \pm 1.5	18.1 \pm 5.5	18.6 \pm 2.3	6.3 \pm 1.2	5.6 \pm 2.1	107 \pm 42	5.0 \pm 2.1
21	6.3	11.4 \pm 0.5	69.5 \pm 2.6	113 \pm 2	50.8 \pm 1.2	28.5 \pm 0.3	6.5 \pm 0.5	8.7 \pm 0.2	138 \pm 49	4.1 \pm 1.6
22	6.1	8.3 \pm 0.4	41.3 \pm 9.8	105 \pm 2	33.0 \pm 0.8	16.2 \pm 0.1	3.7 \pm 0.4	4.1 \pm 0.4	81.9 \pm 39.0	3.8 \pm 1.9
23	6.1	11.8 \pm 1.2	165 \pm 17	199 \pm 5	154 \pm 8	75.7 \pm 1.0	15.9 \pm 0.8	18.8 \pm 0.8	353 \pm 150	4.2 \pm 2.3
24	6.3	11.0 \pm 1.3	82.5 \pm 5.6	173 \pm 4	49.6 \pm 2.6	32.9 \pm 1.4	9.7 \pm 0.5	10.8 \pm 0.6	158 \pm 54	5.2 \pm 1.3
25	6.4	9.1 \pm 0.8	110 \pm 7	141 \pm 21	82.5 \pm 9.9	45.1 \pm 2.6	10.5 \pm 1.7	12.0 \pm 1.7	218 \pm 65	3.8 \pm 2.3
26	6.3	23.1 \pm 1.8	18.9 \pm 6.6	155 \pm 5	BLQ	1.0 \pm 0.0	4.4 \pm 0.7	2.1 \pm 0.3	26.0 \pm 6.7	4.3 \pm 1.8
27	6.6	5.3 \pm 0.2	147 \pm 7	151 \pm 8	209 \pm 9	32.2 \pm 0.7	22.5 \pm 1.1	26.0 \pm 1.2	426 \pm 36	115 \pm 6.9
Mean \pm SD	6.3 \pm 0.2	10.6 \pm 5.7	94.1 \pm 45.8	144 \pm 41	90.4 \pm 61.8	34.8 \pm 21.0	10.6 \pm 5.8	11.4 \pm 7.1	204 \pm 126	21.0 \pm 37.3

BLQ, below the limit of quantification (0.15 $\mu\text{g/mL}$). Total catechins = EGCG + EGC + GCG + ECG + EC.

be the same as that reported in tea leaves. Nakagawa *et al.*⁽²⁰⁾ has reported the similar observation. The same study also showed that co-administration of CA and EGCG (0.7 - 3 mg CA and 1.6 mg EGCG per kg of body weight) increased the bioavailability of EGCG in humans by 36 - 60%⁽²⁰⁾. These results suggest that CA in green tea drink may enhance the bioavailability of catechins especially that of the EGCG.

III. Concentrations of Sugars in Tea Drinks

Because tea catechins and alkaloids (caffeine and theobromine) contribute to the bitterness of tea drinks, some commercial tea drinks are supplemented with sugars to mask the bitter taste. Table 5 shows the concentrations of glucose, sucrose, and fructose in the tea drinks. Sucrose was the main sweetener in these tea drinks. All black tea and some green and Oolong tea drinks were supplemented with sucrose. In addition to sucrose, Sample 12 was also found supplemented with glucose (18.3 g/L) and fructose (22.4 g/L). It was demonstrated that sugar-sweetened beverages are a significant contributor to weight gain and can lead to increased risks of type 2 diabetes mellitus and cardiovascular diseases^(14,15). This result implies that sugar-sweetened tea drinks may counteract the beneficial effects of unsweetened tea drinks in reducing the risk of these diseases. Therefore, sugar-free tea drinks are considered to provide greater health benefits.

IV. Correlation Coefficient Analyses between Tea Components and Antioxidant Activities

The principal hypothesis is that the putative benefits of tea are associated with the strong free radical-scavenging and/or antioxidant properties of their polyphenols, especially catechins⁽⁵⁾. For all tea drinks, there was a strong correlation between the antioxidant activities and total phenolics (DPPH: $r = 0.80$; ABTS: $r = 0.88$) (Table 6). Regression correlation coefficients also showed important contributions of the contents of EGC (DPPH: $r = 0.82$; ABTS: $r = 0.87$), GCG (DPPH: $r = 0.69$; ABTS: $r = 0.80$), ECG (DPPH: $r = 0.59$; ABTS: $r = 0.65$), and EC (DPPH: $r = 0.80$; ABTS: $r = 0.76$). The free radical scavenging activities also had high correlations with total catechins (DPPH: $r = 0.79$; ABTS: $r = 0.87$). The ascorbic acid contents only weakly influenced the antioxidant potencies of the tea drinks (DPPH: $r = 0.28$; ABTS: $r = 0.3$). GA showed the greatest contribution to the free radical scavenging activities only in black tea drinks (DPPH: $r = 0.95$; ABTS: $r = 0.77$). In sugar-sweetened tea drinks, the total sugar content showed weak or negative correlations with the free radical scavenging activities (DPPH: $r = -0.22$; ABTS: $r = -0.16$) (data not shown). These results indicated that the greatest contribution of phenolic compounds to the free radical scavenging activities of the tea drinks, especially green and Oolong tea drinks, were from EGC, EC, ECG, and GCG. It was interesting to note that CA also had a high correlation with free radical scavenging activities in green tea drinks (DPPH: $r = 0.74$; ABTS: $r = 0.78$). However, in this study, CA at concentrations of up to 200 $\mu\text{g/mL}$ had no

significant effect on the inhibition of DPPH and ABTS radicals ($< 3\%$) (data not shown). This result can be explained by the fact that tea catechins and CA are extracted out simultaneously by hot water during the brewing process.

Catechins were found to effectively scavenge various radical species (i.e. hydroxyl and DPPH radicals) although their respective scavenging abilities tended to differ somewhat according to the type of the radical species tested⁽²⁴⁾. The important structure features of catechins for scavenging the DPPH radical are *ortho*-trihydroxy group in the B ring and

Table 5. Concentrations of glucose, sucrose, fructose, and total sugars of tea drinks examined in this study (g/L)^a

	D-Glucose	Sucrose	D-Fructose	Total sugars ^a
Green tea drink				
1	— ^b	1.3 \pm 0.1	—	1.3 \pm 0.1
2	—	1.5 \pm 0.1	—	1.5 \pm 0.1
3	—	—	—	—
4	—	—	—	—
5	—	4.1 \pm 0.0	1.0 \pm 0.0	5.1 \pm 0.1
6	—	—	—	—
7	—	65.5 \pm 0.5	—	65.5 \pm 0.5
8	—	39.2 \pm 0.1	—	39.2 \pm 0.1
9	—	—	—	—
10	—	3.4 \pm 0.2	—	3.4 \pm 0.2
Black tea drink				
11	—	41.0 \pm 0.2	—	41.2 \pm 0.0
12	18.1 \pm 0.3	43.7 \pm 2.4	19.4 \pm 2.1	81.2 \pm 4.8
13	—	38.1 \pm 2.6	—	38.2 \pm 2.6
14	—	59.5 \pm 4.8	—	59.5 \pm 4.8
15	—	68.6 \pm 4.4	—	68.6 \pm 4.4
16	—	60.3 \pm 6.9	—	60.3 \pm 6.9
17	—	70.2 \pm 1.7	—	70.2 \pm 1.7
Oolong tea drink				
18	—	—	—	—
19	—	3.0 \pm 0.8	0.9 \pm 0.1	3.9 \pm 1.1
20	—	—	—	—
21	—	31.0 \pm 4.1	—	31.1 \pm 4.1
22	—	—	—	—
23	—	—	—	—
24	—	—	—	—
25	—	—	—	—
26	—	35.6 \pm 4.2	—	35.6 \pm 4.2
27	—	35.7 \pm 4.3	—	35.7 \pm 4.3

^a Total sugars, glucose + sucrose + fructose.

^b The sugar concentration was below the quantification limit of 0.5 g/L.

Table 6. Linear correlation coefficient (*r*) between tea components and antioxidant activities of tea drinks examined in this study

	Total phenolics	Ascorbic acid	GA	CA	EGCG	EGC	GCG	ECG	EC	EGCG + GCG	Total catechins
Total tea drinks											
DPPH	0.80	0.28	-0.02	0.13	-0.10	0.82	0.69	0.59	0.80	0.69	0.79
ABTS	0.88	0.30	-0.08	0.16	-0.07	0.87	0.80	0.65	0.76	0.79	0.87
Green tea drinks											
DPPH	0.67	0.20	0.00	0.74	0.39	0.50	0.52	0.17	0.31	0.56	0.64
ABTS	0.74	0.33	0.00	0.78	0.44	0.60	0.75	0.18	0.26	0.67	0.75
Black tea drinks											
DPPH	0.48	-0.07	0.95	0.06	0.00	0.90	0.71	-0.13	0.95	0.00	0.50
ABTS	0.90	-0.37	0.77	0.28	0.18	0.78	0.70	0.00	0.58	0.08	0.69
Oolong tea drinks											
DPPH	0.74	0.20	-0.05	0.32	0.00	0.89	0.66	0.78	0.80	0.78	0.84
ABTS	0.86	0.17	-0.04	0.34	0.00	0.90	0.68	0.73	0.78	0.79	0.84

In the linear correlation analysis, tea components were regarded as X and % of DPPH and ABTS radical scavenging activities as Y. GA, gallic acid; EGCG, epigallocatechin gallate; EGC, epigallocatechin gallate; GCG, gallic acid; ECG, epigallocatechin gallate; EC, epicatechin; Total catechins = EGCG + EGC + GCG + ECG + EC.

a galloyl moiety at the 3 position and, therefore, the DPPH scavenging abilities of EGC, ECG and EGCG are stronger than other catechins⁽²⁵⁾. EGCG appeared to be the most effective scavenger regardless of the radical species^(24,26). In contrast to a previous study which indicated that EGCG in green tea has a high correlation ($r > 0.8$) with the DPPH-scavenging activity⁽²⁷⁾, in this study, EGCG in green tea drinks showed a relatively weak influence on the free radical scavenging activities (DPPH: $r = 0.39$; ABTS: $r = 0.44$). One of the possible explanations for this observation is the partial conversion of EGCG to GCG in tea drinks. This speculation might be supported by the observation that EGCG + GCG also had high correlations with the antioxidant activities in green (DPPH: $r = 0.69$; ABTS: $r = 0.79$) and Oolong tea drinks (DPPH: $r = 0.78$; ABTS: $r = 0.79$). Moreover, similar inhibitory potencies (IC_{50} values) of EGCG and GCG against DPPH (EGCG: 3 $\mu\text{g/mL}$; GCG: 3 $\mu\text{g/mL}$) and ABTS (EGCG: 2.5 $\mu\text{g/mL}$; GCG: 2.3 $\mu\text{g/mL}$) radicals were noted (Table 2). Similar antioxidant potencies between EGCG and GCG for both human low-density lipoprotein oxidation and DPPH free radical assays were reported by Xu *et al.*⁽²²⁾. These results indicated that in green and Oolong tea drinks, the epimerization from EGCG to GCG might not cause any loss of their antioxidant capacities, but could reduce the weight of EGCG as a major contributor to antioxidant activity. On the other hand, the green tea drinks have higher concentration of EGC than the Oolong and black tea drinks (Table 4); however, the lower correlation between the free radical scavenging activities and the EGC level was found in green tea drinks than in the Oolong and black tea drinks (Table 6). These results suggested that components other than catechins, such as ascorbic acid and unknown phenolic compounds, in green tea drinks may also contribute to the antioxidant activity and reduce the correlation between the free radical scavenging activities and the EGC level. Nevertheless, EGCG and EGC, due to their high radical scavenging abilities and high levels, were suggested as the major antioxidants in green tea drinks.

It is noteworthy that the total phenolic contents and the inhibiting activities against DPPH and ABTS free radicals of green tea drinks were approximately 2-fold higher than that of black tea drinks (Table 2). However, the extent of differences on total catechins between green and black tea drinks was much greater (8.8-fold) than the differences on the inhibiting activities against DPPH and ABTS radicals (Table 4). This result clearly indicates that components other than catechins in black tea drinks may also contribute to the free radical scavenging activities. It was demonstrated that theaflavins, the fermentation product of catechins in black tea, and catechins in green tea are equally effective antioxidants, suggesting that the fermentation process may not greatly reduce the antioxidant properties⁽²⁸⁾. In this study, our data also showed similar free radical scavenging activities between GA and catechins (Table 2). Therefore, even the total phenolics and free radical scavenging activities of black tea drinks were lower than those of green tea drinks, their antioxidant potentials cannot be ignored.

Tea polyphenols, especially catechins, were

demonstrated to prevent oxidative modifications of cellular lipids, proteins, and nucleic acids by multidirectional antioxidant actions⁽²⁹⁾. Thus regular consumption of tea drinks may provide beneficial effects in reducing the risk of chronic diseases such as cardiovascular disease and cancers. Assuming an average tea consumption in Taiwan to be one to two bottles (approximately 600 mL per bottle) per day, the calculated mean values of total phenolics intake from green tea drinks in Taiwan would be approximately 0.29-0.58 g/day, which is close to one half of that of a typical American diet (1 g/day)⁽³⁰⁾. It is suggested that commercial tea drinks represent an excellent source of phenolic compounds that exhibit important antioxidant activities.

In summary, total phenolics, total catechins, and EGC levels of tea drinks were major contributors to the scavenging abilities against DPPH and ABTS radicals. Green tea drinks have higher free radical scavenging activity than black and Oolong tea drinks may be related to their high levels of total phenolics and catechins. Our data also show that commercial tea drinks, especially the green tea drink, can be an important dietary source of polyphenols with antioxidant activity. In Taiwan, this is the first study concerning the major components (catechins, gallic acid, total phenols, caffeine, sugars, ascorbic acid) of three types of commercial tea drinks and their correlations to free radical scavenging activity. Because canned and bottled tea drinks are becoming more popular worldwide, the results of the present study may provide health information to consumers. Our data suggest that more data of tea components such as catechins, caffeine and sucrose should be provided on the labels of commercial tea drinks.

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REFERENCES

1. Harbowy, M. E. and Balentine, D. A. 1997. Tea chemistry. *Crit. Rev. Food Sci. Nutr.* 16: 415-480.
2. Zuo, Y., Chen, H. and Deng, Y. 2002. Simultaneous determination of catechins, caffeine and gallic acids in green, oolong, black and pu-erh teas using HPLC with a photodiode array detector. *Talanta* 16: 307-316.
3. Hollman, P. C., Feskens, E. J. and Katan, M. B. 1999. Tea flavonols in cardiovascular disease and cancer epidemiology. *Proc. Soc. Exp. Biol. Med.* 220: 198-202.
4. Cabrera, C., Artacho, R. and Gimenez, R. 2006. Beneficial effects of green tea-a review. *J. Am. Coll. Nutr.* 25: 79-99.
5. Benzie, I. F. and Szeto, Y. T. 1999. Total antioxidant capacity of teas by the ferric reducing/antioxidant power assay. *J. Agric. Food Chem.* 47: 633-636.
6. Cabrera, C., Gimenez, R. and Lopez, M. C. 2003. Determination of tea components with antioxidant activity. *J. Agric. Food Chem.* 51: 4427-4435.
7. Manach, C., Williamson, G., Morand, C., Scalbert, A. and Remesy, C. 2005. Bioavailability and bioefficacy of polyphenols in humans. I. Review of 97 bioavailability studies. *Am. J. Clin. Nutr.* 81: 230S-242S.
8. Ho, C. T., Lee, C. Y. and Huang, M. T. 1994. Phenolic compounds in Food in their effect on health. II. Antioxidant and Cancer prevention. *Free Radic. Biol. Med.* 16: 669-670.
9. Nehlig, A., Daval, J. L. and Debry, G. 1992. Caffeine and the central nervous system: mechanisms of action, biochemical, metabolic and psychostimulant effects. *Brain Res. Rev.* 17: 139-170.
10. Seto, R., Nakamura, H., Nanjo, F. and Hara, Y. 1998. Preparation of epimers of tea catechins by heat treatment. *Biosci. Biotechnol. Biochem.* 61: 1434-1439.
11. Chen, Z., Zhu, Q. Y., Tsang, D. and Huang, Y. 2001. Degradation of green tea catechins in tea drinks. *J. Agric. Food Chem.* 49: 477-482.
12. Chen, Z. Y., Zhu, Q. Y., Wong, Y. F., Zhang, Z. H. and Chung, H. Y. 1998. Stabilizing effect of ascorbic acid on green tea catechins. *J. Agric. Food Chem.* 46: 2512-2516.
13. Chen Y. L. and Lin, S. Y. 2005. The stability of the solution of catechins. *Taiwan. Tea Res. Bull.* 24: 77-90.
14. Schulze, M. B., Manson, J. E., Ludwig, D. S., Colditz, G. A., Stampfer, M. J., Willett, W. C. and Hu, F. B. 2004. Sugar-sweetened beverages, weight gain, and incidence of type 2 diabetes in young and middle-aged women. *JAMA.* 292: 927-934.
15. Malik, V. S., Popkin, B. M., Bray, G. A., Despres, J. P. and Hu, F. B. 2010. Sugar-sweetened beverages, obesity, type 2 diabetes mellitus, and cardiovascular disease risk. *Circulation* 121: 1356-1364.
16. Singleton, V. L. and Rossi, J. A. 1965. Colorimetry of total phenolics with phosphomolybdic-phosphotungstic acid reagents. *Am. J. Enol. Vitic.* 16: 144-158.
17. de Quiros, A. R.-B., Fernandez-Arias, M. and Lopez-Hernandez, J. A. 2009. screening method for the determination of ascorbic acid in fruit juices and soft drinks. *Food Chem.* 116: 509-512.
18. Zhao, J., Liu, T., Ma, L., Yan, M., Gu, Z., Huang, Y., Xu, F. and Zhao, Y. 2009. Antioxidant and preventive effects of extract from *Nymphaea candida* flower on in vitro immunological liver injury of rat primary hepatocyte cultures. *Evid Based Complement Alternat Med.* 1-8. February 5, doi:10.1093/ecam/nep003.
19. Ozgen, M., Reese, R. N., Tulio, A. Z., Jr., Scheerens, J. C. and Miller, A. R. 2006. Modified 2,2'-azino-bis-3-ethylbenzothiazoline-6-sulfonic acid (abts) method to measure antioxidant capacity of Selected small fruits and comparison to ferric reducing antioxidant power (FRAP) and 2,2'-diphenyl-1-picrylhydrazyl (DPPH) methods. *J. Agric. Food Chem.* 54: 1151-1157.
20. Nakagawa, K., Nakayama, K., Nakamura, M., Sookwong, P., Tsuduki, T., Niino, H., Kimura, F. and Miyazawa, T. 2009. Effects of co-administration of

- tea epigallocatechin-3-gallate (EGCG) and caffeine on absorption and metabolism of EGCG in humans. *Biosci. Biotechnol. Biochem.* 73: 2014-2017.
21. Tsai, Y. S., Liu, S. L., Wang, H. F. and Ou, S. M. 2004. Comparison of catechins contents and antioxidant activity of green teas from Taiwan major tea cultural cultivars. *Taiwan. Tea Res. Bull.* 23: 115-132.
 22. Xu, J. Z., Yeung, S. Y., Chang, Q., Huang, Y. and Chen, Z. Y. 2004. Comparison of antioxidant activity and bioavailability of tea epicatechins with their epimers. *Br. J. Nutr.* 91: 873-881.
 23. Fernández, P. L., Pablos, F., Martín, M. J. and González, A. G. 2002. Study of catechin and xanthine tea profiles as geographical tracers. *J. Agric. Food Chem.* 50: 1833-1839.
 24. Nanjo, F., Mori, M., Goto, K., Hara, Y. 1999. Radical scavenging activity of tea catechins and their related compounds. *Biosci. Biotechnol. Biochem.* 63:1621-1623.
 25. Nanjo, F., Goto, K., Seto, R., Suzuki, M., Sakai, M., Hara, Y. 1996. Scavenging effects of tea catechins and their derivatives on 1,1-diphenyl-2-picrylhydrazyl radical. *Free Radic. Biol. Med.* 21: 895-902.
 26. Gardner, P. T., McPhail, D. B., Duthie, G. G. 1998. Electron spin resonance spectroscopic assessment of the antioxidant potential of teas in aqueous and organic media. *J. Sci. Food Agric.* 76: 257-262.
 27. Saito, S. T., Gosmann, G., Saffi, J., Presser, M., Richter, M. F. and Bergold, A. M. 2007. Characterization of the constituents and antioxidant activity of Brazilian green tea (*Camellia sinensis* var. *assamica* IAC-259 Cultivar) extracts. *J. Agric. Food Chem.* 55: 9409-9414.
 28. Leung, L. K., Su, Y., Chen, R., Zhang, Z., Huang, Y. and Chen, Z. Y. 2001. Theaflavins in black tea and catechins in green tea are equally effective antioxidants. *J. Nutr.* 131: 2248-2251.
 29. Higdon, J. V. and Frei, B. 2003. Tea catechins and polyphenols: health effects, metabolism, and antioxidant functions. *Crit. Rev. Food. Sci. Nutr.* 43: 89-143.
 30. Kühnau J. 1976. The flavonoids. A class of semi-essential food components: their role in human nutrition. *World Rev. Nutr. Diet.* 24: 117-191.